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AP Chemistry

1m magnesium chloride. 1m glucose. 1m sodium chloride. 2m glucose. Correct answer: 1m magnesium chloride. Explanation: The equation for boiling point elevation is $\Delta T_b = iK_b m$. Since all of the solutions are aqueous, we do not need to consider the boiling point elevation constant (K_b) when comparing the solutions.

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Nitrates, acetates, chlorates, and perchlorates are soluble. Chlorides, bromides, and iodides are generally soluble, unless they contain copper, silver, lead, or mercury. Sulfates are soluble, unless they contain barium, lead, silver, strontium, or calcium. Most silver salts are insoluble unless listed in rule #2.

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A solution that contains the maximum amount of dissolved solute.

A solution that contains less than the maximum amount of dissolved solute.

AP Chemistry: Solutions - Practice Test Questions ...

Exercise 7 Calculating the VP of a Solution Containing Two

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Liquids. A solution is prepared by mixing 5.81 g acetone (C_3H_6O , molar mass = 58.1 g/mol) and 11.9 g chloroform ($CHCl_3$, molar mass = 119.4 g/mol). At $35^\circ C$, this solution has a total vapor pressure of 260. torr.

AP* Chemistry PROPERTIES OF SOLUTIONS

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AP Chemistry: Properties of Solutions Lecture Outline 13.1 The Solution Process
A solution is a homogeneous mixture of solute and solvent. Solutions may be gases, liquids, or solids. Each substance present is a component of the solution. The solvent is the component present in the largest amount. The other components are the solutes.

AP Chemistry: Properties of Solutions Lecture Outline 13.1 ...

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015 - Solutions In this video Paul Andersen explains the important properties of solutions. A solution can be either a solid, liquid or gas but it must be ho...

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A solution containing 3.23 grams of an unknown compound dissolved in 100.0 grams of water freezes at -0.97°C . The solution

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does not conduct electricity. Calculate the molecular weight of the compound.

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