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1.1

Pure Substances
and Mixtures,
Elements \u0026amp;
Compounds,

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of Matter,
Chemistry
Examples,

3.2.1/3.2.2 Chapter

Describe the
differences
between
elements,
compounds and
mixtures. ~~States
of Matter :~~

~~Solid Liquid Gas~~

~~10 Amazing~~

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~~Change~~ Water Changes in

the State of

Matter - Science

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Solid to Liquid,

Liquid to Gas,

Gas to Liquid

Changes of

States of Matter

Conservation of

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physical change.

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will be formed.

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SOLUTIONS MANUAL

35. Aqueous solutions of potassium iodide and silver

nitrate are mixed, forming the precipitate silver iodide.

chemical

equation: $KI(aq)$

$AgNO_3(aq) \rightarrow KNO_3$

$(aq) AgI(s)$

complete ionic

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Equation: $K(aq)$

$I(aq)$ 32 Ag

(aq) $NO_3(aq)$ 0

$K(aq)$ $NO_3(aq)$

$AgI(s)$ Chapter

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Chemical Reaction

Chemical

Reactions

Solution.

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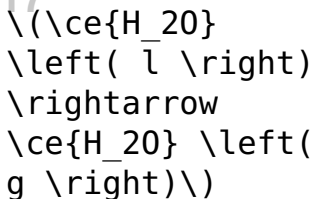
boiling and

melting are

physical

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changes. When water boils, no bonds are broken or formed. The change could be written:



Chemical: The dark grey nail changes color to

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form an orange
flaky substance
(the rust); this
must be a
chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

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12 SOLUTIONS
MANUAL 30.

Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid

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phase directly
to the vapor
phase. During
evaporation,
particles in a
liquid gain
enough

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