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in Model 2. a. Which color of light

has the most energy? b. Which
color of light has the least

energy? 12. Does a gas discharge
tube filled with boron emit the

same wavelengths of light as a
tube filled with hydrogen? Use

evidence from Model 2 to support
your answer. 13.

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a. Which of the electron

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transitions involves the most energy? $n=6$ to $n=2$, violet light, 410 nm. b. Explain why this transition involves the most energy based on your understanding of the attractive forces between the electrons and protons in the atom. As the electron moves toward the nucleus of the atom there is a stronger attraction. This releases

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