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Formula Pogil Science Division

The empirical formula for a compound is C_2H_5 and its relative formula mass is 58.0. Deduce its molecular formula. Deduce its molecular formula. (Relative atomic masses: C = 12.0, H = 1.0)

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Empirical Formula Pogil Science Division Bellevue empirical formula. 2. Divide the molecular mass by the empirical mass.

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3. Multiply all of the subscripts in the empirical formula by the resulting factor. Example Problem: What is the molecular formula for a material that is 92.2% C and 7.80% H by mass? The material has a molar mass of 52.0 g/mol. 1. Molecular mass /

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