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11 REVIEW Gases

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ANSWER Answer

the following

questions in the

space provided.

1. _____ The

molar mass of a

gas at STP is

the density of

that gas (a)

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Chemistry by
the mass of 1
mol. (c)

multiplied by
22.4 L. (b)

divided by the
mass of 1 mol.

(d) divided by
22.4 L. 2.

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manual online or
by storing it on

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Three of the primary components of air are carbon dioxide, nitrogen, and oxygen. In a sample containing a mixture of only these gases at

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exactly 1 atm,
the partial
pressures of
carbon dioxide
and nitrogen are
given as P_{CO_2}
 $= 0.285$ torr and
 $P_{\text{N}_2} = 593.525$
torr. What is
the partial
pressure of
oxygen?

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Convert a
Section 3 Gases
pressure of
Answer Key

0.200 atm to the
following units:

152 a. mm Hg

20.3 b. kPa 6.

When an
explosive like
TNT is
detonated, a
mixture of gases
at high
temperature is

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Answer Key

Created. Suppose that gas X has a pressure of 50 atm, gas Y has a pressure of 20 atm, and gas Z has a pressure of 10 atm. 80 atm a. What is the total ...

*mc06se cFMsr i-
vi*

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ANSWER Answer

the following
questions in the
space provided.

1. List the
following gases
in order of
increasing rate
of effusion.

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ANSWER Answer

the following

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Questions in the
space provided.

1. _____ The
molar mass of a
gas at STP is
the density of
that gas (a)
multiplied by
the mass of 1
mol. (c)
multiplied by
22.4 L. (b)
divided by the
mass of 1 mol.

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(d) divided ...

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the space
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temperature and
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